

LETTER TO THE EDITOR

Theoretical and isotopic infrared absorption investigations of nitrogen–oxygen defects in silicon

R Jones†, C Ewels†, J Goss†, J Miroz‡, P Deák‡, S Öberg§ and F Berg Rasmussen||

† Department of Physics, University of Exeter, Exeter EX4 4QL, UK

‡ Department of Atomic Physics, Technical University of Budapest, Budafoki út 8, H-1111 Budapest, Hungary

§ Department of Mathematics, University of Luleå, Luleå, S-95187, Sweden

|| Institute of Physics and Astronomy, University of Aarhus, DK-8000, Århus C, Denmark

Received 18 August 1994, accepted for publication 22 September 1994

Abstract. The vibrational spectroscopy of NNO defects in Si introduced by ^{16}O , ^{14}N and ^{15}N ion implantation is studied, and especially the N-isotopic shifts of the localized vibrational modes. These investigations show that the local modes of the three impurity atoms comprising the defect are only weakly coupled dynamically. *Ab initio* cluster calculations of the local mode frequencies of the defect are performed. Several models are investigated, and the model consisting of a bridging O atom adjacent to the N pair defect accounts for its dynamic properties.

Nitrogen is commonly used as an inert atmosphere for processing Si but it has important effects on the material. For example, N impurities can pin dislocations [1, 2] and form electrically active defects such as the substitutional deep donor [3, 4]. There have been suggestions that nitrogen–oxygen complexes can form shallow donors [4–6], although whether N does form an essential component of these defects is controversial [7, 8].

The dominant N defect in O-free Si consists of a pair of N_i atoms (figure 1) and has C_{2h} symmetry [4, 9]. The N-pair is stable until about 800 °C [4]. In N- and O-implanted FZ-Si [10], or in CZ-Si grown in a N_2 atmosphere [11, 12], NNO complexes are formed by annealing between 400 and 700 °C. Under these conditions, the intensity of the localized vibrational modes (LVMs) due to the pair is progressively reduced and three other IR-absorption lines at 1026, 996 and 801 cm^{-1} (room temperature) become increasingly prominent. The intensities of these lines are correlated with each other [12] and they are believed to be due to NNO complexes. This is because O_i atoms become mobile around 400 °C and become trapped by N-pair defects, which are stable at this temperature. Beyond ≈ 700 °C, these complexes break up with the emission of O_i , leaving the N-pair. Cooling allows the NNO complexes to reform. This dissociation and reformation is reversible.

There are no reports of isotopic effects on the LVMs of the NNO defect even though this information is of the greatest importance in deducing its structure. To remedy this, we have carried out infrared absorption studies on NNO defects created by ion implantation.

Float-zone n-type Si samples cut from a wafer were implanted at room temperature with ^{16}O and with either ^{14}N , ^{15}N or both N-isotopes. The samples had a typical size of $10 \times 10 \times 0.3 \text{ mm}^3$ and were polished and implanted on both sides. The energies (table 1) and doses were chosen to obtain a nearly uniform O and N profile from 0.21 to 0.82 μm below the surface with a concentration of $6.5 \times 10^{19} \text{ cm}^{-3}$ for both O and N. All

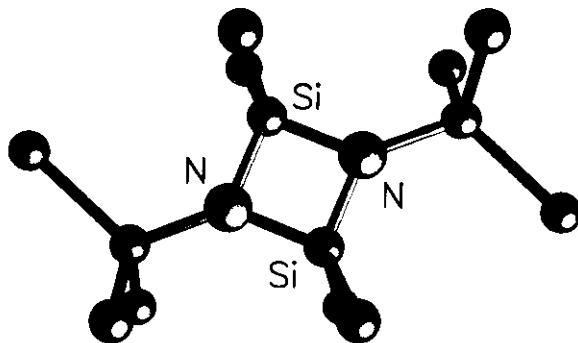


Figure 1. The N-pair defect.

Table 1. Energies (in keV) of implanted ions.

^{16}O	^{14}N	^{15}N
103	90	96
171	150	161
257	225	241
371	325	348
457	400	429

samples were annealed under flowing N_2 ambient for an hour in the temperature range from 500 to 650 °C and were subsequently etched by HF to remove surface oxides. Infrared absorption measurements (figure 2) were carried out at 77 K using a Nicolet System 800 Fourier transform infrared spectrometer. The number of scans was typically 1000 and a resolution of 6 cm^{-1} was chosen to avoid fringes in the spectra due to multiple reflection of the beam in the sample. The LVMs due to the NNO defect are given in table 2.

The ^{14}N frequencies agree with the earlier studies [11–13], allowing for their shift ($\approx 4\text{ cm}^{-1}$) between low and room temperature. All the modes shift with ^{15}N , showing that at least one N atom is involved in the defect. Unexpectedly, the peak in the 1030 cm^{-1} LVM is shifted by around 3 cm^{-1} with ^{15}N , and this suggests that this mode is almost entirely due to Si–O stretch. This lies between the LVMs for O_i (1136 cm^{-1} [14]) and O_s (836 cm^{-1} [15]), suggesting that O_i bridges a dilated Si bond near the N-pair. The modes at 999 and 805 cm^{-1} are slightly above those of an isolated N-pair (967 and 770 cm^{-1}) and show a similar shift with ^{15}N (25 and 19 cm^{-1} compared with 26 and 17 cm^{-1} for the N-pair), suggesting that the pair remains intact but with compressed bonds due to the nearby O_i . No additional LVMs are observed in the mixed implantation, implying that the N atoms are no longer dynamically coupled but vibrate independently of each other. They must also be inequivalent atoms because they possess distinct LVMs.

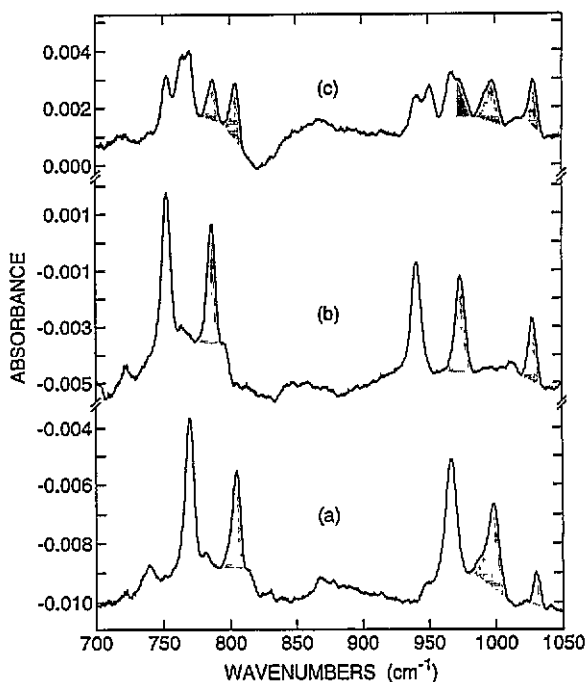


Figure 2. Infrared absorption intensity (547 °C, 1 h): (a) ^{14}N , (b) ^{15}N , (c) ^{14}N – ^{15}N . The shaded bands are those of NNO defects. The other main peaks are due to N_2 pairs.

This implies that O lies closer to one N atom than the other.

Calculations on the N-pair defect [9] showed that N_i dilated the Si(1)–Si(5) bond (figure 1) by 4%, making it a prime site for O_i . The other Si bonds bordering the pair along [100] are compressed, making these unlikely sites for the O_i . Previous investigations on the trapping of O_i at dislocation cores showed a marked preference for O to bridge dilated Si bonds [16].

Two models of the NNO defect are shown in figure 3. They differ in the position of the O atom. In the first, the O atom bridges the nearest Si–Si bond to the pair, and in the second, O bridges the second nearest Si–Si bond along [011].

Table 2. Calculated and observed LVMs, (in cm^{-1}), due to N–N–O complexes in Si.

$^{14}\text{N } ^{14}\text{N } ^{16}\text{O}$	$^{15}\text{N } ^{15}\text{N } ^{16}\text{O}$	$^{14}\text{N } ^{15}\text{N } ^{16}\text{O}$	$^{15}\text{N } ^{14}\text{N } ^{16}\text{O}$	$^{14}\text{N } ^{14}\text{N } ^{18}\text{O}$
Calculated				
1070.4	1039.6	1069.8	1040.3	1070.3
861.1	860.1	860.7	860.4	823.9
808.2	788.1	788.2	808.1	804.5
723.9	705.5	708.3	722.2	723.9
671.5	655.8	669.5	656.7	670.6
Displaced Si atom				
962.9	962.7	962.8	962.8	949.2
948.9	920.1	948.0	921.3	918.9
813.3	792.6	792.7	813.0	813.2
727.0	708.2	714.0	723.9	727.0
688.1	672.3	683.5	674.2	687.4
Observed				
1030	1027		1028	
999	974		999, 974	
805	786		805, 786	

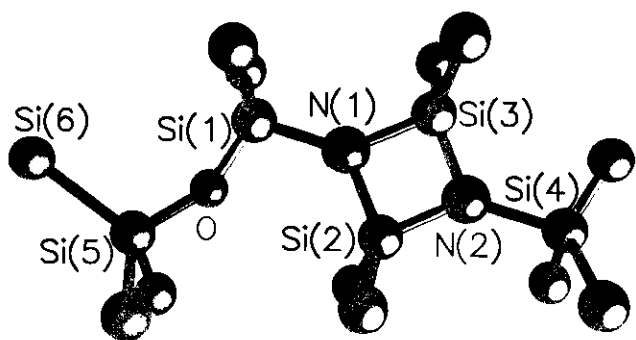


Figure 3. Structure of the N-N-O defect.

LDF cluster calculations [17] were carried out on a 134 atom cluster $N_2OSi_7H_{60}$ containing the defect as shown in figure 1. The method has been described before [17] and gives Si-Si bonds to within 1.5% and successfully describes both the vibrational modes of O_i [18] as well as its activation energy for diffusion [19].

The self-consistent energy and forces on all the atoms were found, and the whole cluster relaxed by a conjugate gradient algorithm. The entries in the dynamical matrix of the cluster between the impurities and their neighbours were determined directly, and the other entries were found from a Musgrave-Pople potential derived earlier [17].

The vibrational modes of the structure are given in table 2. The mode at 861 cm^{-1} is due to Si-O stretch because its frequency is almost unaffected by ^{15}N substitution. The two modes at 1070 and 808 cm^{-1} are related to N(1) and N(2), and the mixed isotopic results show very little coupling between them. This explains why no new modes are introduced in this case, since the nearby O_i makes the two N atoms inequivalent.

Thus, although the modes are in the correct region and display the correct isotopic behaviour, their ordering is incorrect: the O mode should lie above the two highest N modes. This ordering and their positions are, however, very sensitive to the position of the common Si atom, Si(1) (figure 1). If this is displaced by 0.07 \AA along $[0\bar{1}\bar{1}]$, decreasing the Si(1)-O and increasing the N(1)-Si(1) lengths by 0.05 and 0.07 \AA respectively, then the O and N modes are within 70 cm^{-1} of the observed frequencies (table 2). Once again the atoms are decoupled. The highest O-mode is simply an asymmetric stretch in the Si-O bonds. There is a symmetric stretch mode at lower frequency. The mode at 949 cm^{-1} is due to N(1)-Si(1) stretch whereas the 813 cm^{-1} mode represents N(2)-Si(3) stretch. The other modes involve motion of both N atoms. The two N-related modes shift by 28 and 19 cm^{-1} with ^{15}N , in excellent agreement with the observed shifts of 25 and 19 cm^{-1} . The changes in bond lengths caused by the shift of Si(1) and the discrepancies between the calculated and observed modes lie within the errors of the calculation.

We also investigated three other models of the defect. Moving the O_i away from the N-pair to lie between Si(5) and Si(6) raised the cluster energy by 1.2 eV . Neither moving the O_i to lie between N(1) and Si(1) in figure 1

nor using O to bridge the dilated bond in the Humble N-pair model [9] gave modes in the right region or caused them to shift correctly when ^{14}N is replaced by ^{15}N .

Our conclusion is that the structure in figure 1 represents the NNO defect. The three atoms of the defect are only weakly dynamically coupled because of their inequivalence. The O atom is bound to the defect because the N-pair strains the lattice. The dissociation energy of the NNO complex is then approximately the energy necessary for O to migrate to the barrier top, starting from O lying between Si(6) and Si(5) and ending between Si(5) and Si(1). Taking the first figure to be that for O_i diffusion, 2.5 eV [20], and the second one as calculated here, 1.2 eV , we find this dissociation energy to be 3.7 eV . The equilibrium concentration of NNO defects would then be

$$\frac{[N_2][O_i]}{N_s} \exp(1.2\text{ eV}/kT)$$

where N_s is the density of bond-centre sites for O_i . The formation energy of NNO from N_2 and O defects is estimated to be around 1.2 eV , which gives a concentration of NNO equal to that of N_2 below $800\text{ }^\circ\text{C}$ if $[N_2] = [O_i] = 10^{-6}N_s = 10^{17}\text{ cm}^{-3}$.

Wagner *et al* [11] report two weak extra LVMs at 810 and 1018 cm^{-1} . We suggest that these lines are due not to NNO but possibly to NNO_2 defects.

We remark that, as the trapping of O by N pairs has been attributed to the strain field surrounding the pair, then this mechanism would be also be expected to operate for other impurities. H and F would behave like O. However, C would be expected to substitute for Si atoms compressed by the pairs whereas Ge and substitutional transition metals would replace Si atoms subject to tensile stress. This would lead to $\langle 011 \rangle$ -oriented aggregates.

Finally, we note that the NNO defect has no deep or shallow donor levels. This suggests that, if shallow thermal donors [5, 6] are N-related, they must involve larger numbers of O or N atoms, or possibly include Si interstitials [21].

SÖ thanks the Swedish National Research Council, FBR thanks the Danish Research Foundation through ACAP (Århus), PD and JM thank the Hungarian National Scientific Research Fund (OTKA 2744) and RJ thanks the British Council, for financial support.

References

- [1] Sumino K, Yonenaga I, Imai M and Abe T 1983 *J. Appl. Phys.* **54** 5016
- [2] Heggie M I H, Jones R and Umerski A 1993 *Phys. Status Solidi a* **138** 383
- [3] Brower K L 1973 *Phys. Rev. B* **26** 6040
- [4] Stein H J 1985 *Oxygen, Carbon, Hydrogen and Nitrogen in Crystalline Silicon (Mater. Res. Soc. Symp. Proc.)* ed J C Mikkelsen Jr *et al* (Pittsburgh, PA: MRS) vol 59 p 523

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- [5] Suezawa M, Sumino K, Harada H and Abe T 1988 *Japan. J. Appl. Phys.* **27** 62
- [6] Hara A, Aoki M, Koizuka M and Fukuda T 1994 *J. Appl. Phys.* **75** 2929
- [7] Griffin J A, Navarro H, Weber J and Genzel L 1986 *Mater. Sci. Forum* **10–12** 997
- [8] Heijmink Liesert B J, Gregorkiewicz T and Ammerlaan C A J 1993 *Phys. Rev. B* **47** 7005
- [9] Jones R, Öberg S, Berg Rasmussen F and Bech Nielsen B 1994 *Phys. Rev. Lett.* **72** 1882
- [10] Stein H J 1985 *Mater. Res. Soc. Symp. Proc.* ed N M Johnson *et al* (Pittsburgh, PA: MRS) vol 46 p 287
- [11] Wagner P, Oeder R and Zulehner W 1988 *Appl. Phys. A* **46** 73
- [12] Qi M W, Tan S S, Zhu B, Cai P X, Gu W F, Xu X M, Shi T S, Que D L and Li L B 1991 *J. Appl. Phys.* **69** 3775
- [13] Qi M W, Shi T S, Tan S S, Zhu B, Cai P X, Liu L Q, Que D L and Li L B 1992 *Mater. Sci. Forum* **83–87** 263
- [14] Kaiser W, Frisch H L and Reiss H 1958 *Phys. Rev.* **112** 1546
- [15] Bosomworth D R, Hayes W, Spray A R L and Watkins G D 1970 *Proc. R. Soc. A* **317** 133
- [16] Umerski A and Jones R 1993 *Phil. Mag. A* **67** 905
- [17] Jones R 1992 *Phil. Trans. R. Soc. A* **341** 351
- [18] Jones R, Umerski A and Öberg S 1992 *Phys. Rev. B* **45** 11321
- [19] Jones R, Öberg S and Umerski A 1990 *Mater. Sci. Forum* **65–66** 287
- [20] Newman R C and Jones R 1994 *Oxygen in Silicon* (in the series *Semiconductors and Semimetals*) ed F Shimura (New York: Academic) ch VI, to be published
- [21] Deák P, Snyder L C and Corbett J W 1992 *Phys. Rev. B* **45** 11612